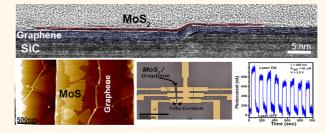
Direct Synthesis of van der Waals Solids

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ABSTRACT The stacking of two-dimensional layered materials, such as semiconducting transition metal dichalcogenides (TMDs), insulating hexagonal boron nitride (hBN), and semimetallic graphene, has been theorized to produce tunable electronic and optoelectronic properties. Here we demonstrate the direct growth of MoS₂, WSe₂, and hBN on epitaxial graphene to form large-area van der Waals heterostructures. We reveal that the properties of the underlying graphene dictate properties of the heterostructures, where strain, wrinkling, and defects



on the surface of graphene act as nucleation centers for lateral growth of the overlayer. Additionally, we show that the direct synthesis of TMDs on epitaxial graphene exhibits atomically sharp interfaces. Finally, we demonstrate that direct growth of MoS₂ on epitaxial graphene can lead to a 10³ improvement in photoresponse compared to MoS₂ alone.

raphene is considered the foundation of exciting new science in twodimensional layered materials,¹ but it is only the "tip of the iceberg". Novel device designs necessarily require additional high-quality film as either the barrier or active layer. Recently, hexagonal boron nitride (hBN) has attracted attention as a gate dielectric or substrate material for integration with graphene-based electronics as a gate dielectric or substrate material because its sp²-hybridized bonding and weak interlayer van der Waals bonds result in a pristine interface.² This also leads to a decreased density of absorbed impurities that act as Coulombic scattering centers when designing novel layered heterostructures.^{3–5} Additionally, two-dimensional dichalcogenide-based materials are of significant interest for their finite band gaps ranging from 3.5 eV for GaS⁶ to <1 eV for MoTe₂ and WTe₂.⁷ More specifically, transition metal dichalcogenides (TMDs) such as molybdenum disulfide (MoS₂), tungsten disulfide (WS₂), and tungsten diselenide (WSe₂), have gained momentum in recent years due to their applications in a variety of

electronic and optoelectronic applications.³ This is also complemented by the possibility to tune the energy band gap of TMDs from 0.8 to 2.1 eV through heterogeneous integration, thus producing entirely novel electronic and optoelectronic materials not yet synthesized.⁸ These unique properties make TMDs promising candidates for highperformance, low-cost energy materials for use in flexible electronics, photovoltaics,⁹ and energy storage.^{10,11} Development of electronically tunable van der Waals solids (vdW) must start with high-quality substrate materials. Graphene and graphite are excellent templates for the growth of bilayer AB-stacked graphene,¹² topological insulators,¹³ and other 2D materials such as hBN and MoS₂.^{14–18} To date, progress in the development of vdW heterostructures has led to a variety of new phenomena.^{16,18} However, these vdW structures are primarily fabricated via mechanical exfoliation using polymer membranes and micromanipulators to stack the individual 2D crystals.⁴ The process of mechanical exfoliation, while often useful for demonstration purposes, can lead to interface contamination.¹⁹ These defects

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and adsorbates buried at the interface of the 2D crystals undoubtedly diminish the quality and performance of devices.¹⁹ Therefore, the development of a growth technique to assemble these systems during synthesis is required for large-area, high-quality vdW solids. Several groups have demonstrated direct growth of bilayer vdW solids (two dissimilar layers) as the building block for further heterointegration. Liu et al.¹⁴ used a two-step ex situ process to grow chemical vapor deposition (CVD) graphene followed by synthesis of hBN on the CVD graphene. Similarly, Shi et al.¹⁵ utilized CVD graphene as the growth template to grow MoS₂ by flowing (NH₄)₂MoS₄ precursor. While these are significant advances in the pursuit of vdW solids, the use of CVD graphene on Cu requires sophisticated methods of Cu etching and transferring to avoid pinholes, tears, or surface contamination in the heterogeneous structures. This may limit the applicability of heterostructures based on CVD graphene simply due to variation in electronic properties due to polymeric contamination, mechanical strains, and substrate/vdW solid interface imperfections. In this work, we utilize epitaxial graphene (EG) on 6H-SiC as the growth template for direct growth of MoS₂, WSe₂, and hBN. Quasi-free-standing epitaxial graphene (QFEG), hydrogen-treated EG,²⁰ is also utilized as a growing template and compared to EG to understand the impact of the EG buffer layer. Epitaxial graphene is utilized because it provides several technological advantages: (1) graphene is already on an insulating substrate, requiring no transfer processes; (2) the interface between graphene and SiC is pristine and tailorable;²⁰ (3) the surface is free of polymeric and other contaminants found in transferred CVD graphene;²¹ and (4) epitaxial graphene is typically quite robust under standard device fabrication processes.²² However, there are also challenges to utilization of epitaxial graphene (uniform thickness over large areas, steps in the SiC surface) that must also be considered. Here, we utilize atomic force microscopy (AFM), Raman spectroscopy, transmission electron microscopy (TEM), X-ray photoelectron spectroscopy (XPS), and photocurrent measurements to elucidate the impact of epitaxial graphene properties on the resulting MoS₂/EG, hBN/EG, and WSe₂/EG heterostructures.

RESULTS AND DISCUSSION

The utilization of epitaxial graphene provides a near perfect template for synthesis of vdW solids due to its lack of dangling bonds, chemical inertness, and ability to remain intact under high stress. Additionally, many of the 2D layered materials are isostructural, belonging to the symmetry group $P6_3/mmc.^{23}$ However, on the basis of the lattice parameters of graphene, hBN, MoS₂, and WSe₂ (2.461, 2.50, 3.16, and 3.28 Å, respectively),²³

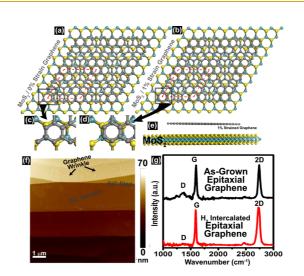


Figure 1. (a–e) DFT model of $MoS_2/unstrained$ graphene and $MoS_2/1\%$ strained graphene predicts that residual strain in epitaxial graphene may enhance the structural symmetry in TMD/graphene heterostructures. (f) AFM image of as-grown epitaxial graphene demonstrating a smooth surface along with 5–10 nm steps in the SiC substrate. Wrinkles in graphene grown on SiC (0001) appear due to the CTE mismatch between graphene and SiC. (g) Raman spectra of as-grown epitaxial graphene and H₂-treated (hydrogenated) epitaxial graphene. Following hydrogenation, the D peak in EG is nearly eliminated by buffer layer passivation.

one may not expect epitaxial growth to proceed for transition metal dichalcogenides on graphene because a significant lattice mismatch exists. While this mismatch can be highly detrimental in 3D semiconductor epitaxy, there is a significant relaxation in the required lattice matching when growth proceeds via vdW interaction.²⁴ Density functional theory (DFT) clearly indicates poor commensurability between strain-free graphene and MoS₂ (Figure 1a,c), where the red dashed box is a guide for the eye, illustrating that the closest lattice match occurs at 3 and 4 unit cells (3-4) for MoS₂ and graphene, respectively. Figure 1c further illustrates the poor commensurability of the MoS₂/graphene (and other TMD/graphene combinations) heterostructures at the (3-4) matchup. However, the weak vdW interaction provides a route for "epitaxy" to occur in these structures. Ongoing DFT simulations, with van der Waals corrections as implemented in ONETEP,²⁵ have already shown homogeneous symmetrical reconstruction of the layered structures by inducing large strain on the layer of graphene at the local density approximation level of exchange and correlation functional.²⁶ Additionally, DFT indicates that the residual strain in epitaxial graphene (typical for graphene on SiC)²⁷ could impact epitaxy of the TMD by providing a route for improved commensurability (Figure 1b,d,e). Synthesis of highquality epitaxial graphene was prepared by using an in situ hydrogen etch, followed by Si sublimation from the Si face of semi-insulating 6H-SiC (II-VI, Inc.) at 1700 °C, 100 Torr.²² Some EG samples were further

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exposed to molecular hydrogen (hydrogenation) at 1050 °C, 600 Torr, for 60 min to passivate the graphene/SiC (0001) buffer layer-referred as guasifree-standing epitaxial graphene.²⁰ Hydrogenation typically results in the partial relaxation of residual strain in epitaxial graphene, which is evident in the red-shifted 2D Raman peak of the QFEG (Figure 1g). Raman and transmission electron microscopy confirm a uniform, continuous top layer of EG extending over the SiC $(1\overline{1}0n)$ step edge and onto SiC (0001) terraces, with few-layer EG existing on the step edge after the growth (Supporting Information Figure S1).²² The QFEG exhibits atomically smooth surfaces on the SiC terrace, as well as atomic-scale wrinkles that result from the coefficient of thermal expansion (CTE) mismatch that exists between EG and SiC (Figure 1f). This provides an ideal platform for understanding the interaction of heterolayers and the impact surface defects such as wrinkling, thickness, and surface potential variation have on the ability to form pristine vdW solids.²⁰ In addition to transforming the buffer into an additional graphene layer, Raman indicates that the quality of QFEG is significantly improved compared to EG, as the D peak is suppressed (Figure 1g). EG and QFEG are utilized as the base templates for direct integration with MoS₂.

The nucleation and growth of MoS₂ is strongly influenced by the characteristics of the underlying graphene. Heterolayers were synthesized via vaporphase reaction of sublimed MoO_{3-x} and sulfur powders in a horizontal furnace, using both EG and QFEG as the growth template.²⁸ Using EG as the template, there is a distinct pattern for multilayer, pyramidalshaped MoS₂ to form on the SiC (0001) terrace, while smooth mono- and bilayer MoS_2 forms at the SiC(110*n*) plane and extends outward (Figure 2a). On the other hand, under the same synthesis conditions, monolayer MoS₂ dominates the surface coverage on QFEG (Figure 2b) and is complemented by multilayered MoS_2 islands nucleating on wrinkles, SiC (110n) step edge, and at graphene defects. The variation in nucleation and growth of MoS₂ on EG and QFEG is due to the difference in graphene strain, thickness,²⁹⁻³¹ and potentially buffer layer—all of which have significant impact on the chemical reactivity of graphene.³¹ As-grown EG is thinner, exhibits 1% residual strain, and has a buffer layer that is partially covalently bound to the SiC substrate.³⁰ However, in the case of multilayer EG, where the impact of the buffer layer is expected to be significantly reduced due to increased EG thickness, MoS₂ tends to grow laterally rather than vertically. In the case of QFEG, the hydrogenation process passivates the buffer layer and decouples the graphene layers from the underlying SiC (Figure S2a). This results in some strain relief of the graphene (~0.3-0.5% residual strain), increases the layer thickness, and removes the EG buffer.

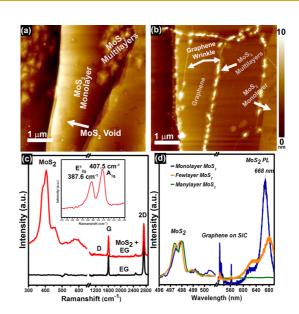


Figure 2. Atomic layers of MoS₂ grown on (a) EG and (b) QFEG demonstrating a clear difference in nucleation and growth for the two graphene template, which is attributed to the presence of enhanced residual strain and a buffer layer in EG compared to QFEG. In the case of QFEG, MoS₂ primarily nucleates on wrinkles and SiC step edges. (c) Raman spectra of MoS₂/EG and EG before the direct growth. Raman indicates that synthesis of MoS₂ does not induce additional defects in the EG or QFEG. Inset: E_{2g} and A_{1g} of the MoS₂ Raman features. (d) Photoluminescence from multilayer MoS₂ to monolayered MoS₂ after the growth, demonstrating that high-quality mono-, bi-, and multilayer MoS₂ is possible on epitaxial graphene.

This suggests that the chemical reactivity of EG is higher than QFEG due to strain and the presence of a buffer layer in EG, which results in a high density of TMD nucleation sites on EG.²⁰

The MoS₂/EG(QFEG) heterogeneous structures exhibit high-quality structural, chemical, and interfacial properties (Figure 2c,d). Upon synthesis, the samples display MoS_2 peaks at 386 and 408 cm⁻¹ (E_{2g}^{1} and A_{1g} peaks, respectively)³² and epitaxial graphene peaks at 1590 cm⁻¹ (G peak) and 2740 cm⁻¹ (2D peak)³⁰ in the Raman spectra (Figure 2c and inset). The intense E_{2q}^{1} and A_{1q} vibration modes of the MoS₂ Raman spectra indicate that the MoS₂ is of high quality.³² The E_{2q}^{1} and A_{1q} peak spacing is also an indicator of layer thickness in TMDs^{32,33} and is found to be 20.0 cm⁻¹ for monolayer MoS₂ on epitaxial graphene. This value is 2 cm^{-1} larger than that in exfoliated monolayer MoS₂ and is due to the stiffening of the A_{1g} mode with *in situ* synthesis and CTE mismatch.³³ Photoluminescence (PL) from the MoS₂/EG(QFEG) heterostructures also provides evidence that the films are of high crystalline quality and can range from mono- to few-layer (Figure 2d) under current growth conditions.³⁴ Importantly, we note that Raman spectroscopy also confirms the direct synthesis of MoS₂ on EG and QFEG does not impact the quality of the underlying graphene (Figure 2c, no "D" peak). The element analysis on MoS₂/EG and MoS₂/QFEG via

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XPS also confirms a proper stoichiometric ratio between Mo and S of 2, with no carbon bonding. The Mo 3d exhibits two peaks at 229.3 and 232.5 eV (Figure S2b), which is attributed to the doublet $3d_{5/2}$ and $3d_{3/2}$, respectively. The peaks corresponding to sulfur $2p_{1/2}$ and $2p_{3/2}$ orbitals in bonded sulfide are observed at 164 and 162.5 eV (Figure S2c). It should be noted that the resulting shapes of Mo and S in XPS are very similar, and thus only the MoS₂/QFEG heterostructure case is presented.

Following nucleation, there are distinct growth morphology differences between EG and QFEG. As noted in Figure 2a and Figure S3a, there exists a high density of MoS₂ islands on the terrace of EG, which are hypothesized to grow in a manner similar to that described by Stranski-Krastanov.³⁵ Here, the MoS₂ films form via a combination of layer-by-layer growth and 3D island growth, which is similar to synthesis of graphene on hBN.³⁶ The source of this islanding phenomenon may be related to the vertical propagation of a defect that initiates in the graphene layer (vacancy, strain-induced high chemical reactivity site, etc.; we later discuss the impact of graphene quality on MoS₂), which forms an additional nucleation site for adlayers to form and grow laterally outward from the central peak of the island (Figure 2b and Figure S3b). This phenomenon is also present in the QFEG case, although fewer isolated islands are identified on these samples where pristine QFEG exists. Rather, the QFEG wrinkles are more reactive due to the curved sp² π -bonds, which again induce localized strain and modification of the chemical reactivity of the graphene films.³⁷ The same phenomenon also occurs on QFEG step edges, similar to the synthesis of MoS₂ ribbons on highly oriented pyrolytic graphite (HOPG) surface.³⁸ The grain size of individual MoS₂ crystals on EG ranges from 200 to 1000 nm for the 10 min growth (Figure S4); however, these domains only partially cover the graphene surface. Longer growth times lead to a coalescence of MoS₂ domains into high-quality MoS₂ sheets. The MoS₂ domain size is increased by reducing the nucleation sites and prolonging the time for the growth. Surface potential force microscopy (SPFM) measurements provide evidence that the step edge region exhibits a 60 mV lower surface potential than that of graphene on the SiC terrace (Figure S5a,b), confirming that graphene on the SiC step edge is more reactive. The impact of MoS₂ synthesis on the quality of QFEG is correlated with the measured layer thickness of the MoS₂ overlayer. Evident in Figure 3, there are regions of high $I_{\rm D}/I_{\rm G}$ ratios in the Raman spectra (Figure 3a,d), which closely match the presence of wrinkles in the graphene film, as well as areas of "thick" MoS₂ (Figure 3c,e). While wrinkles are present in as-grown QFEG (Figure 1f), the I_D/I_G is <0.05, indicating that high-quality QFEG exists prior to MoS₂ synthesis. It is not until after the synthesis of MoS₂ that significant degradation occurs at regions of the QFEG at wrinkle locations, which also correlates with broadening of the

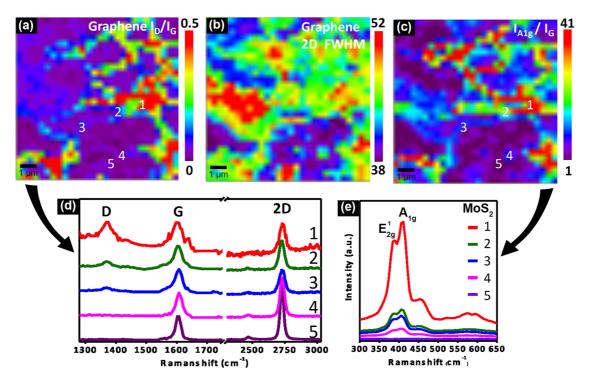


Figure 3. (a) Raman ratio map of QFEG I_D/I_G indicates that the synthesis of MoS₂ can induce defects in the underlying graphene layers at nucleation sites (*i.e.*, wrinkles, steps). (b) Corresponding 2D fwhm map of QFEG shows that broadening of the graphene 2D width can correlate the MoS₂ thickness. (c) Raman peak ratio map of $I_{A_{19}}$ of MoS₂ to I_G of QFEG ratio indicates the MoS₂ distribution on QFEG, where higher ratios indicate "thicker" MoS₂. Comparing (a,c), there is a clear correlation between defects and MoS₂ thickness at QFEG wrinkles. (d,e) Raman spectra of numbered locations in the Raman map presented the MoS₂ and QFEG (a–c).

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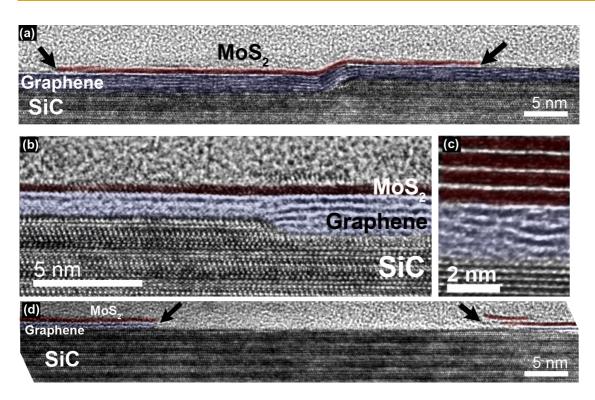


Figure 4. (a) Cross-sectional HRTEM of $MoS_2/QFEG$ demonstrating the nucleation and subsequent lateral growth of MoS_2 on a SiC step edge covered with QFEG. In this case, the graphene thickness is consistent across the step, resulting in bending of the graphene and variation in strain of the EG layer. On the other hand, when the top graphene layer remains flat (b), the MoS_2 grows without regard of the changing morphology below. When the underlying graphene is defective (c), additional MoS_2 adlayers are present, which indicate that defects in the graphene can produce this MoS_2 . Finally, MoS_2 nucleation and growth is promoted by the presence of graphene, while often it is found to be absent on bare SiC.

2D peak (Figure 3b). Correlating I_D/I_G (Figure 3a) with the ratio of the A_{1g} peak in MoS₂ and G peak in QFEG (Figure 3c) clearly indicates that the defect level is higher along QFEG wrinkles and in locations where the Raman spectra indicate bulk MoS₂ (Figure 3c,e). Interestingly, every MoS₂ island associated with a wrinkle appears to be bisected by the wrinkle (Figure 2b), suggesting that the nucleation of the MoS₂ occurs at the peak of the wrinkle and is followed by lateral growth of the layer. This scenario is highly plausible considering that the wrinkle apex is of highest stress-and thus the highest chemical reactivity.³¹ It has been reported that chemical functionalization on graphene is energetically preferable to happen at the site with higher chemical reactivity due to a lower formation energy.³¹ Thus, we speculate that the formation energy of MoS₂ nucleation is lower at the location with high chemical reactivity.

The direct synthesis of TMD on QFEG results in a pristine heterointerface. Transition electron microscopy cross-sectional micrographs confirm that monolayer MoS_2 nucleates on the $(1\overline{1}0n)$ step edges and subsequently extends onto the (0001) terrace in regions where the thickness of graphene does not vary over the $(1\overline{1}0n)/(0001)$ conjunction (Figure 4a). This suggests that the $(1\overline{1}0n)$ step edges may play a critical role in nucleation, which is consistent with the observation in AFM images of $MoS_2/EG(QFEG)$ in Figure 2a,b.

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Furthermore, the MoS₂ layer appears to be "blind" to thickness variations in the underlying graphene when there are no defects in the top layer of the graphene (Figure 4b). Apparently, in the case where additional layers of graphene are formed in a manner as to maintain a flat surface profile, the graphene appears to shield the influence from the SiC morphology. On the other hand, when defective graphene is present at the surface of the graphene layer, there is almost always multilayered MoS₂ present (Figure 3c). Additionally, guite often it is found that MoS₂ nucleation and growth does not proceed on the SiC (0001) terrace when graphene is not present. Figure 4d demonstrates that the MoS₂ growth ends at the terminations of graphene, which suggests that graphene may serve as the catalytic layer for MoS₂ nucleation and growth. The similar case has been reported in the growth of MoS₂ on CVD graphene/Cu.¹⁵ However, the MoS₂ grown directly on catalytic metals has not been reported.

Photoconductivity has been demonstrated on singlelayer MoS₂, but the photoconductivity must be improved to be competitive to current state-of-the-art materials. Yin *et al.*³⁹ demonstrate a photoresponsivity of ~0.42 mA/W with at a photon intensity of 80 μ W and drain bias of 1 V, while 7.5 mA/W is achieved at $V_g = 50$ V. This was improved upon by Zhang *et al.*⁴⁰ with the mechanically stacked heterostructures CVD MoS₂/CVD graphene, where the photoresponsivity

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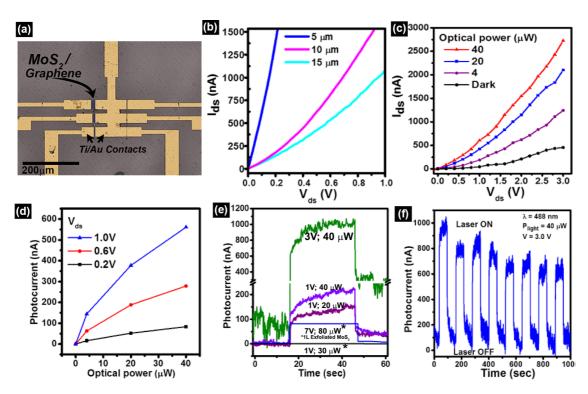


Figure 5. (a) Optical image of the fabricated MoS₂/graphene photosensor. (b) Room temperature electrical characteristic of a two-terminal MoS₂/QFEG photosensor shows the drain current (I_{ds}) vs voltage (V_{ds}) at different channel lengths (5, 10, and 15 μ m). The output characteristics of the MoS₂/QFEG photosensor (c,d) when illuminated with 488 nm photon at increasing illuminating laser power and gate voltage. (e) Photoswitching behavior of the photosensor at different laser power (P_{light}) and V_{ds} compared to current literature.³⁹ (f) Transient measurement of the photosensor at $V_{ds} = 3.0$ V and $P_{light} = 40 \ \mu$ W, demonstrating fast, stable response.

reached more than 10^7 A/W with the electrical gating effect. We also investigate photoconductivity and responsivity using two-terminal MoS₂/QFEG devices without back-gating fabricated via standard photolithography process to demonstrate the superiority of direct synthesis heterostructures.^{41,42} The channel resistance is found to increase from hundreds of ohms to $\sim 10^6 \Omega$ as the channel length increases from 5 to 15 μ m (Figure 5a). Interestingly, the drain current is similar to previous reports for MoS₂³⁹ and much lower than epitaxial graphene devices (Figure S7a),²² indicating that the Ti/Au metal contacts have not contacted the underlying graphene, and that transport is dominated by MoS₂ in the channel. This is encouraging for the development of lateral TMD devices on graphene because the graphene has not shorted the device. This phenomenon is likely a result of the high interlayer resistance in vdW solids,43 which provides a large potential barrier for vertical transport and results in charge being confined to the top MoS₂ layers when the ohmic metallization does not contact the underlying graphene. In the case where pinholes exist in the MoS₂ layer under the ohmic metallization, we find that current transport is dominated by graphene, and I-Vcurves become highly linear with channel resistances <500 Ω . These devices were not considered for photocurrent measurements. Direct growth of MoS₂ on epitaxial graphene yields significant improvements

in photoconductivity and responsivity. The MoS₂/QFEG photosensor device (Figure 5a) was investigated by exploring photocurrent under various optical powers and bias conditions. Ideally, the generation of photocurrent needs to match the basic condition that the incident photon energy must be greater than the optical energy gap (E_q) of MoS₂. Single-layer MoS₂ is reported to have a band gap of 1.83 eV corresponding to 676 nm in wavelength.³⁴ The MoS₂ from the direct growth in this study shows PL at 668 nm, corresponding to an optical energy gap 1.85 eV (Figure 2d). Under a constant excitation wavelength at 488 nm, and power ranging from 4 to 40 μ W, the MoS₂/QFEG generates a power-dependent photocurrent ranging from 150 to 550 nA at $V_{ds} = 1$ V (Figure 5c,d). This represents a 200fold increase in photocurrent compared to exfoliated MoS₂.³⁹ Photoresponsivity is a crucial parameter when evaluating the performance of a photosensor. It is defined as $R = I_{ph}/P_{light}$, where I_{ph} is photogenerated current and P_{light} is the total incident optical power on the photosensor.^{39,40} In the current MoS₂/EG device with a 15 μ m channel length, V_{ds} = 1 V, and V_{g} = 0 V, we find R = 40 mA/W when $P_{\text{light}} = 40 \,\mu\text{W}$. Previously, the reported R of similarly biased monolayer MoS₂ photosensor with $P_{\text{light}} = 80 \ \mu\text{W}$ was 0.42 mA/W,³⁹ which was 2 times that of a WS₂ photosensor device (0.22 mA/W).⁴⁴ Additionally, using a 514 nm laser, the photoresponsivity is further increased to 79 mA/W

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when $P_{\text{light}} = 40 \,\mu\text{W}$ and $V_{\text{ds}} = 1 \,\text{V}$ (Figure S6). Thus the MoS₂/EG photosensor represents a minimum 100-fold improvement in photoresponsivity and 1000-fold improvement in absolute photoconductivity, compared to exfoliated single-layer MoS₂ (Figure 5e). At $V_{\rm ds}$ = 1 V, the photocurrent increases from 150 to 230 nA when the P_{light} increases from 20 to 40 μ W. The photocurrent further increases to 1000 nA when the V_{ds} increases from 1 to 3 V at constant P_{light} of 40 μ W. Moreover, the stability of this switching behavior is further tested by sequentially turning the illumination laser on (30 s) and off (30 s). Figure 5f shows the photocurrent stability over 1000 s of continuous operation. Correlating material characterization with optoelectronic performance, the improvement can be ascribed to (1) the high crystal quality of MoS₂ grown on QFEG, and the presence of pristine surfaces and atomically sharp interfaces of the MoS₂/QFEG heterostructures; and (2) the underlying QFEG screen charge scattering sites that may be present on the SiC substrate. To deconvolute the source of the photoresponse, we also examine QFEG-only photosensors. These photosensors exhibit negligible response due to photon excitation (Figure S7b), confirming theoretical predictions that the primary photon interaction, and thus photocurrent generation, of MoS₂/QFEG is attributed to the additive MoS₂ layer on the graphene.45

To determine the universality of epitaxial graphene as a template for the synthesis of high-quality van der Waals solids, we further demonstrate the integration of tungsten diselenide (WSe₂) and hexagonal boron nitride on QFEG substrates. Cross-sectional HRTEMs of WSe₂/QFEG/ SiC (Figure 6a) and hBN/QFEG/SiC (Figure 6b) and their corresponding Raman signatures (Figure 6c,d) indicate that the underlying graphene is not damaged by the synthesis process.² Additionally, TEM confirms a clean van der Waals gap between the different layered materials. After the growth of hBN on QFEG, several Raman peaks are present at 1368 cm⁻¹, which is suggested as the combination of the $E_{2q}\xspace$ mode of $B{-}N\xspace$ vibration at 1372 cm⁻¹ and the D peak of graphene at 1360 cm^{-1.14} The very minor peak at 2950 cm^{-1} (corresponding to the graphene Raman D+G mode) indicates that the growth of hBN only nominally increases the defect density. Evident from XPS (Figure 6d-f), CVD-deposited hBN is nearly stoichiometric and does not affect the chemistry of graphene underneath. The high-resolution N1s and B1s core level peaks in Figure 6d,f are at 191.5 and 398.8 eV, respectively, consistent with reference values for bulk hBN.47 Additionally, the C1s peak remains unchanged following hBN synthesis. In the WSe2/QFEG case, the prominent Raman peaks of the E2q/A1q from WSe2 and the G peak at 1597 cm⁻¹ from graphene/SiC indicate the possibility to integrate other TMD layers over the epitaxial graphene without significantly impacting the graphene structure. As a result, the utilization of an epitaxial graphene template may be considered as an excellent

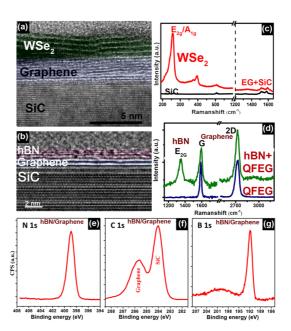


Figure 6. Transmission electron microscopy of (a) WSe₂/ QFEG and (b) hBN/QFEG reveals that the interface within lavers and between heterolavers is pristine, with no observable defects. Additionally, Raman spectra (c,d) of quasifree-standing epitaxial graphene (OFEG), OFEG/hBN, and QFEG/WSe₂ indicate that high-quality van der Waals solids have been grown on QFEG. There is also a small photoluminescence background following hBN synthesis, which has been reported from the CVD-grown hBN thin film. (e-g) High-resolution XPS spectra of B1s, C1s, and N1s core level peaks from hBN grown on QFEG. The peak positions of as-grown B1s and N1s peaks are similar to bulk hBN values, and peak shape indicates no significant contributions from carbon or oxygen bonding in the hBN. The energy loss at 200 eV in (g), which is higher than B1s by 9 eV, is attributed to the $\pi - \pi^*$ interaction in hexagonal phase BN crystals.⁴⁶

candidate for development of a broad range of vertical heterostructures.

CONCLUSIONS

In conclusion, high-quality epitaxial graphene has been synthesized and utilized as the growth template for vdW solids that include MoS₂, WSe₂, and hBN. TMDs are synthesized by gas-phase reaction of MoO₃ and WO₃ and S and Se, while hBN is synthesized on epitaxial graphene by utilizing ammonia borane as the precursor in a tube CVD chamber. This demonstrates that epitaxial graphene may be a universal substrate for a variety of deposition methods and materials. This is due to the presence of residual strain and wrinkles, which play an important role in the nucleation of vdW solids. It has been found that the QFEG template has less MoS₂ nucleated on the terrace than EG template due to the strain relief and elimination of the underlying buffer-both expected to increase chemical reactivity of the graphene. Photosensors based on direct growth MoS₂/QFEG heterostructures exhibit an improvement in photoresponsivity by a minimum of 200-fold, demonstrating the high quality in heterostructures from the direct growth. Finally, we have

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agnanc www.acsnano.org provided strong evidence that epitaxial graphene can be an excellent candidate for building large-area vdW solids that will have extraordinary properties and performance. The two-terminal electrical measurements on synthesized layers here show that the interlayer resistances between the graphene and top heterolayers are high enough and that the stacks are not "shorted" simply by the presence of graphene, rather it will be proper contacting of the individual layers that will be critically important.

EXPERIMENTAL METHODS

Synthesis. Epitaxial graphene is grown on diced SiC wafers via sublimation of silicon from 6H-SiC (0001) at 1700 °C for 15 min under 1 Torr Ar background pressure. Quasi-free-standing graphene is prepared by exposing epitaxial graphene to 600 Torr H_2 at 1050 °C for 120 min to intercalate hydrogen at the graphene/SiC (0001) interface.²² Growth of MoS₂ layers was accomplished using MoO₃ powders (0.1 g) placed in a ceramic crucible located in the center of a 2 in. tube furnace. Sulfur powders are placed in a second ceramic crucible up stream and held at 130 °C during the reaction. The epitaxial graphene/SiC for growing MoS₂ was put at the downstream side. The MoO₃ and S vapors were transported to the epitaxial graphene/SiC substrates by pure Ar flowing gas (Ar = 50 sccm, chamber pressure = 5 Torr). The heating zone was heated to 670 °C at a ramping rate of 15 °C/min.²⁸ Growth of WSe₂ layers on epitaxial graphene were accomplished by first thermally evaporating 5 nm WO₃ on QFEG/SiC. The WO₃/QFEG/SiC was subsequently exposed to selenium vapor by heating pure selenium metal to 500 °C upstream in the tube furnace. This process converts the WO₃ to WSe₂, as discussed elsewhere.⁴⁴ Growth of hBN layers on QFEG was accomplished in a 75 mm diameter horizontal tube furnace via thermal CVD method utilizing ammonia borane (NH₃BH₃) precursor. Solid ammonia borane powder is sublimed at 135 °C and transported into the tube furnace by H₂/Ar carrier gas (5% of total flow rate). Growth occurs at 1075 °C and 250 mTorr in 5 min. After the growth, the ammonia borane carrier gas is turned off and the furnace is allowed to cool to room temperature slowly in a 250 mTorr Ar/H₂ environment.²

Fabrication and Measurement of MoS₂ Photosensor Devices. Twoterminal photosensor devices were fabricated using standard ultraviolet photolithography. The sensors are with various source-drain spacing, which ranges from 1 to 15 μ m. Titanium/ gold (30/100 nm) ohmic contacts were deposited in a similar fashion to our graphene devices, which utilizes an oxygen plasma pretreatment.⁴¹ Photocurrent measurements were carried out at room temperature in ambient conditions and were coupled to a Renishaw micro-Raman spectroscopy with a 488 nm laser. The electrical conduction data were collected with a power source and a Keithley 2400 meter.⁴⁴ The photocurrent measurements at low temperature were acquired using a Lakeshore cryogenic probe station.

Characterization. The as-grown heterostructures were characterized using Raman spectroscopy, AFM, XPS, and TEM. A WITec CRM200 confocal Raman microscope with a 488 nm laser wavelength was utilized for structural characterization. For as-grown epitaxial graphene, the SiC background signal was subtracted using a direct subtraction of the SiC substrate from the spectra.³⁰ A Bruker Dimension with a scan rate of 0.5 Hz was utilized for the AFM measurements. A Kratos Axis Ultra XPS system utilizing an AI K α source with energy of 1486 eV was used for XPS analysis. The TEM cross-sectional samples were made by utilizing a NanoLab dual-beam FIB/SEM system. Protective layers of SiO₂ and Pt were deposited to protect the interesting region during focused ion beam milling. TEM imaging was performed using a JEOL 2100F operated at 200 kV.

Conflict of Interest: The authors declare no competing financial interest.

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Supporting Information Available: AFM, Raman map, SPFM of EG, XPS, and measured photocurrent results under a 514 nm laser of MoS₂/QFEG hybrids and bare QFEG. This material is available free of charge via the Internet at http://pubs.acs.org.

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